

# Exfoliated WS<sub>2</sub> Nanosheets as Photoanodes in Photoelectrochemical Cells

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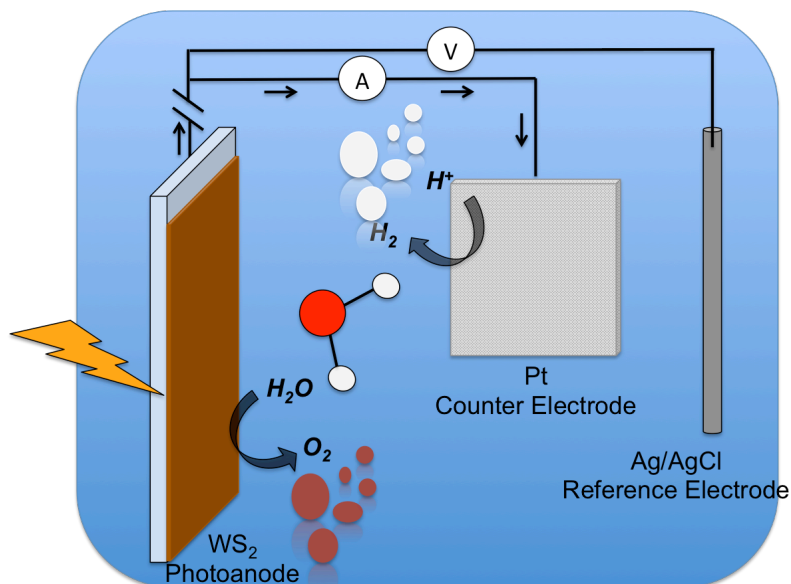
The group 6 of layered transition metal dichalcogenides (TMDs) in their bulk form have long been investigated as suitable materials for applications in photoelectrochemical cells (PECs). Their small band gap enables the absorption of the solar spectrum leading to higher solar conversion efficiency when compared to large band-gap metal oxide semiconductors such as TiO<sub>2</sub>, WO<sub>3</sub> and  $\alpha$ -Fe<sub>2</sub>O<sub>3</sub> widely utilized in PECs for water oxidation. Early works on PECs based on bulk single crystals and polycrystalline films of MoS<sub>2</sub> and WS<sub>2</sub> have shown chemical stability against photocorrosion and solar energy conversion efficiencies for MoS<sub>2</sub> and WSe<sub>2</sub> up to 6% and 17% respectively.[1, 2] However, scalable synthesis of high quality TMDs crystals presented challenges that have hindered further employment in PECs.

A renewed interest in these materials appeared recently owing to their layered nature and the possibility to easily exfoliate them in monolayer sheets because of the weak *van der Waals* forces between the layers. Single layered TMDs of group 6 have now emerged as new semiconducting 2D materials with unique electrical, optical, chemical and mechanical properties complementary to graphene. Their electronic band structure presents a cross over from indirect-to-direct band-gap when the thickness is reduced to single layer.[3] The resulting direct band gap is in the visible range, which can give access to a wide range of optoelectronic applications. WS<sub>2</sub> presents a band gap of 2eV, rendering it suitable for photoassisted water splitting. Further, theoretical studies and early works have shown that the orbital energy levels of WS<sub>2</sub> is favorable for charge transfer for water oxidation.[4]

We employed n-doped monolayered WS<sub>2</sub> platelets as photoanodes in a complete PEC (Figure 1), and WS<sub>2</sub> monolayer films exhibit photocurrent at excitation wavelengths below the direct band gap edge. The material was exfoliated using Lithium intercalation into WS<sub>2</sub> powders followed by exfoliation in water, leading to the formation of a stable aqueous suspension of monolayered WS<sub>2</sub> platelets.[5] Uniform and continuous films of monolayer WS<sub>2</sub> of different thicknesses ranging from 3 to 10nm were fabricated and photoelectrochemically characterised by linear sweep, cyclic voltammetry and chronoamperometry under UV irradiation. WS<sub>2</sub> electrodes exhibited positive photocurrent density values up to 1mA/cm<sup>2</sup> at relative low applied potential versus the Ag/AgCl R.E.. Stability was studied over days enabling possible utilization of such material in practical devices. Chronoamperometry characterizations suggest that improvements in the stacking of WS<sub>2</sub> nanosheets could enhance the transport of photoexcited carriers. These results demonstrate the capability of monolayer WS<sub>2</sub>-based electrodes to be employed in PECs for the production of solar fuels.[6]

## References

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**Figure 1:** Pictorial representation of a PEC based on a nanosheets-WS<sub>2</sub> electrode employed as photoanode. After charge carriers separation, the photogenerated holes in WS<sub>2</sub> can act as oxidising species leading to molecular O<sub>2</sub> evolution from water oxidation, whereas the protons are reduced at a Pt mesh counter electrode.