

# Elastic Properties of Hydrogenated Graphene

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*in collaboration with*

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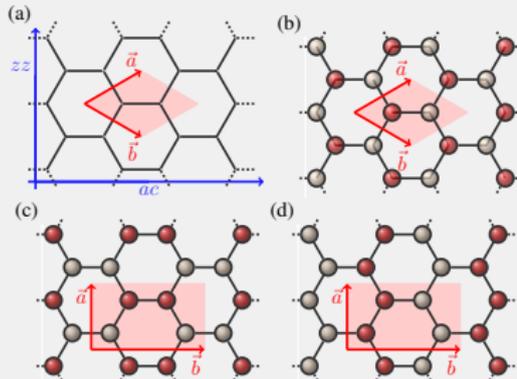


- 1 Introduction
  - Basic definitions
  - Hybridization vs. properties
  - This work
- 2 Theoretical set-up
- 3 Structure and stability
  - Structure
  - Stability
- 4 Calculation of elastic constants
  - Method
  - Validation
  - Second-order stiffness constants (linear elasticity)
- 5 Mechanical behavior under loading
  - Linear regime
  - Nonlinear elasticity
- 6 Conclusions

## Graphane is the hydrogenated form of graphene

- a two-dimensional, periodic, and covalently bonded **hydrocarbon with a C:H ratio of 1**
- hydrogen atoms decorate the carbon honeycomb lattice on **both the top and bottom side**
- any graphane conformer retains **high crystallographic symmetry** (either trigonal or orthorhombic)

### *Possible hydrogen decorations*



**Red:** top hydrogen atoms  
**Grey:** bottom hydrogen atoms

**Shaded areas:** unit cell with lattice vectors

(a): graphene scaffold (full lines) with zigzag (zz) and armchair (ac) directions

(b, c, d): **chair-**, **boat-**, and **washboard-graphane**

- **Graphane**

at first theoretically predicted

*Phys. Rev. B* **75**, 153401 (2007) - *Phys. Rev. B* **77**, 035427 (2008)

eventually grown

*Science* **323**, 610 (2009)

since then, many more papers...

- **Attractive feature**

**hydrogenation:  $sp^2 \rightarrow sp^3$  change in orbital hybridization  $\rightarrow$  new properties**

- **Change in the orbital hybridization**

- ① it has been calculated that graphane is an insulator, with an energy gap as large as  $\sim 3$  eV  
*Phys. Rev. B* **75**, 153401 (2007) - *Phys. Rev. B* **77**, 035427 (2008)
- ② partially H-covered and disordered samples show unlike electronic and phonon properties than graphene  
*Science* **323**, 610 (2009)
- ③ calculated in-plane stiffness and Poisson ratio of graphane are smaller than those of graphene
- ④ yield strain is predicted to vary upon temperature and stoichiometry  
*Topsakal et al., Appl. Phys. Lett.* **96**, 091912 (2010)

**Our perspective:**  $sp^2$ -to- $sp^3$  change in orbital hybridization is expected to cause

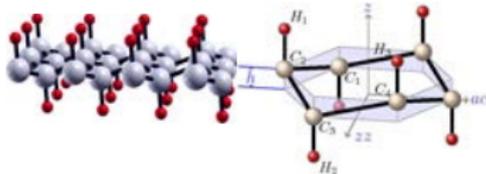
**major variations in the mechanical properties of hydrogenated graphene**

### Highlights

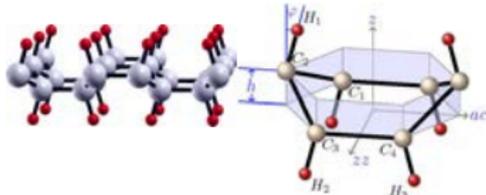
- 1 characterize non-isotropic linear elastic features  
*contrary to graphene: Science 321, 385 (2008) - Phys. Rev. Lett. 102, 235502 (2009)*
- 2 increased or reduced “rigidity” ?  
*linear elastic moduli most likely affected*
- 3 investigate overall mechanical behavior of the systems  
*anisotropic dependence of its response to external loads*

### Approach: blending together atomistic and continuum modeling

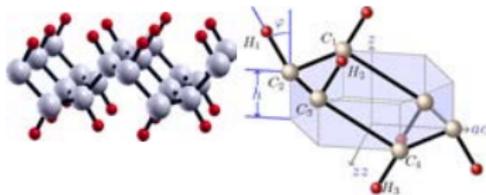
- **Step 0** - Search for equilibrium structures of graphene conformers and establish their stability
- **Step 1** - Calculate 2nd-order  $C_{ij}$  and 3rd-order  $C_{ijk}$  elastic constants
- **Step 2** - Elaborate a continuum description for (anisotropic) elastic moduli
- **Step 3** - Investigate mechanical behavior upon loading



(a) C-graphane



(b) B-graphane



(c) W-graphane

## Graphane conformers have very different structures

### 1 specific H sublattice and different buckling

- C-graphane: H alternate on both sides
- B-graphane: H alternate along ac
- W-graphane: H along  $z$  double rows on both sides

### 2 C-C bond length

- C-graphane and W-graphane: similar to  $sp^3$  systems and much larger than in graphene  
 → bond length  $1.54\text{\AA}$
- B-graphane: two types of C-C bonds
  - ◇ connecting two C atoms lying on opposite sides  
 → bond length  $1.57\text{\AA}$
  - ◇ connecting two C atoms lying on same sides  
 → bond length  $1.54\text{\AA}$

### 3 C-H bond length

typical of any hydrocarbon:  $1.1\text{\AA}$

## Phonons\* and energetics

### 1 phonon branches

- no soft modes (i.e. no instabilities)
- linear  $k$ -dependence for LA and TA modes
- $k^2$ -dependence for flexural (ZA) mode

### 2 anisotropic elastic behavior

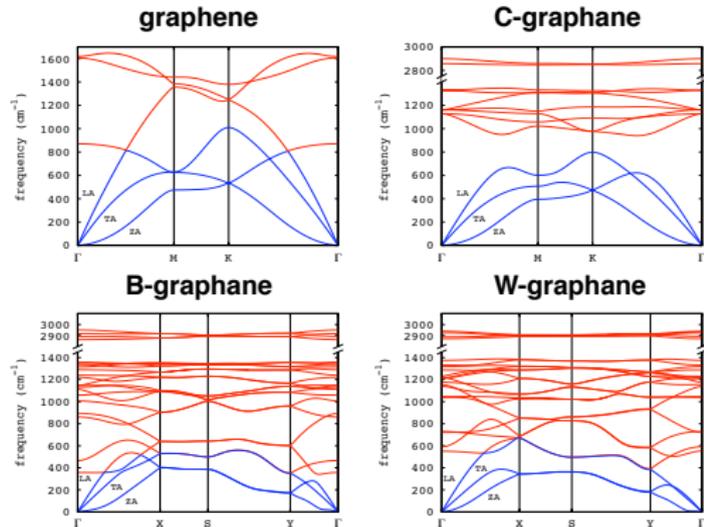
- C-graphane: isotropic speed of sound
- B- and W-graphane: different sound velocities

### 3 energetics

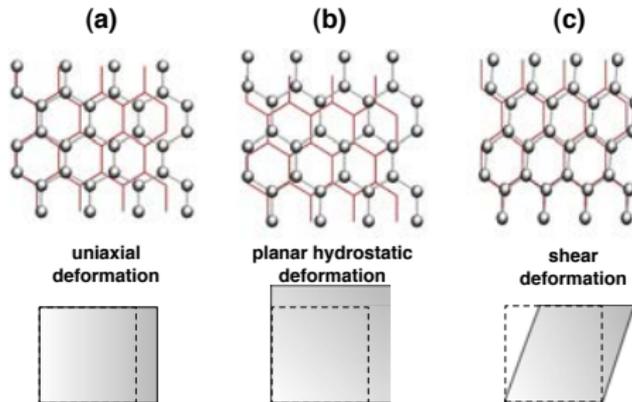
- C-graphane: energetically favored conformer
- W- and B-graphane: higher GS energy  
 0.05 and 0.10 eV per C-H unit
- all conformers thermodynamically accessible,  
 as indeed experimentally guessed

\* Compare well with

- ◇ *Phys. Rev. Lett.* **105**, 037002 (2010) for C-graphane
- ◇ *arXiv:1011.0018v1 [cond-mat.mtrl-sci]*



- Total-energy calculations** performed by Density Functional Theory (DFT)  
QUANTUM ESPRESSO *package* - *J. Phys.: Condens. Matter* **21** 395502 (2009)
  - GGA-like XC functional - Vanderbilt ultrasoft pseudopotential
  - 50Ry energy cutoff - (18x18x3) Monkhorst-Pack grid for BZ sampling
- Atomic positions optimized** by using the quasi-Newton algorithm and PBCs  
*interactions between adjacent atomic sheets hindered by a large spacing > 10 Å*
- Elastic moduli obtained from the (energy) vs. (in-plane strain) curves**  $U = U(\zeta)$



$$U(\zeta) = U_0 + \frac{1}{2}U^{(2)}\zeta^2 + \frac{1}{6}U^{(3)}\zeta^3 + O(\zeta^4)$$

## Benchmark calculations

- energy-vs-strain obtained by in-plane deformations through very many data-points  
*a technically important issue!*
- in-plane deformations defined by a single parameter  $\zeta$
- deformations applied up to 5%

		Young [ $\text{Nm}^{-1}$ ]	Poisson
<b>graphene</b>	this work	344	0.169
	PRB <b>64</b> , 235406 (2001)	345	0.149
	PRB <b>76</b> , 064120 (2007)	350	0.186
<b>C-graphane</b>	this work	246	0.080
	APL, <b>96</b> , 091912 (2010)	243	0.070

	graphene	C-	graphane	W-
		C-	B-	W-
$C_{11}$	354	248	258	280
$C_{22}$			225	121
$C_{12}$	60	20	-1.7	14
$C_{44}$	147(*)	114(*)	93	81

units of  $\text{Nm}^{-1}$       (\*)  $2C_{44} = C_{11} - C_{12}$

### Qualitative trends

- 1  $C_{11} - C_{12}$  difference  
 → **W-graphane the most elastically anisotropic conformer**
- 2  $C_{44}$  modulus  
 → **resistance to a shear deformation decreases monotonically**
- 3  $C_{12}$  modulus (or, similarly, the Poisson ratio)  
 → much smaller than in pristine graphene  
 → **lateral contraction upon extension affected by the new hybridization**
- 4 **B-conformer characterized by a negative  $C_{12}$  value**

- ① **C-graphane**: trigonal symmetry (i.e. elastically isotropic as hexagonal graphene)

$$U_{trigo} = \frac{1}{2}C_{11} (\epsilon_{xx}^2 + \epsilon_{yy}^2 + 2\epsilon_{xy}^2) + C_{12} (\epsilon_{xx}\epsilon_{yy} - \epsilon_{xy}^2)$$

- ② **B- and W-conformers**: orthorhombic symmetry (i.e. anisotropic linear elastic)

$$U_{ortho} = \frac{1}{2}C_{11}\epsilon_{xx}^2 + \frac{1}{2}C_{22}\epsilon_{yy}^2 + C_{12}\epsilon_{xx}\epsilon_{yy} + 2C_{44}\epsilon_{xy}^2$$

- ③ **constitutive in-plane stress-strain equation** are straightforwardly

$$\hat{T} = \frac{\partial U}{\partial \hat{\epsilon}} \rightarrow \begin{cases} T_{xx} & = & C_{11}\epsilon_{xx} + C_{12}\epsilon_{yy} \\ T_{yy} & = & C_{22}\epsilon_{yy} + C_{12}\epsilon_{xx} \\ T_{xy} & = & 2C_{44}\epsilon_{xy} \end{cases}$$

#### Procedure to get elastic moduli

- **apply an axial tension**  $\sigma$  along the arbitrary direction  $\vec{n} = \cos \theta \vec{e}_x + \sin \theta \vec{e}_y$   
 →  $\theta$  is the angle between applied tension and zigzag direction
- **get stress components**:  $T_{xx} = \sigma \cos^2 \theta$      $T_{yy} = \sigma \cos \theta \sin \theta$      $T_{yy} = \sigma \sin^2 \theta$
- invert the constitutive equation → get **longitudinal strain**  $\epsilon_l$     **transverse strain**  $\epsilon_t$
- **elastic moduli**:  $E_{\vec{n}} = \frac{\sigma}{\epsilon_l}$      $\nu_{\vec{n}} = -\frac{\epsilon_t}{\epsilon_l}$

## 1 Orthogonal symmetry

- Young modulus

$$E_{\vec{n}} = \frac{\Delta}{C_{11}s^4 + C_{22}c^4 + \left(\frac{\Delta}{C_{44}} - 2C_{12}\right)c^2s^2}$$

- Poisson ratio

$$\nu_{\vec{n}} = -\frac{\left(C_{11} + C_{22} - \frac{\Delta}{C_{44}}\right)c^2s^2 - C_{12}(c^4 + s^4)}{C_{11}s^4 + C_{22}c^4 + \left(\frac{\Delta}{C_{44}} - 2C_{12}\right)c^2s^2}$$

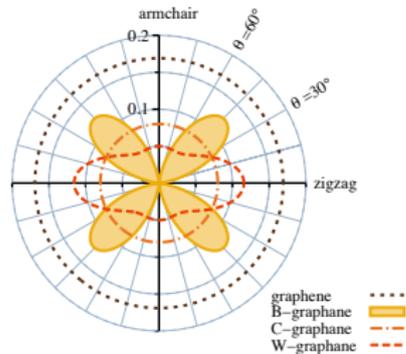
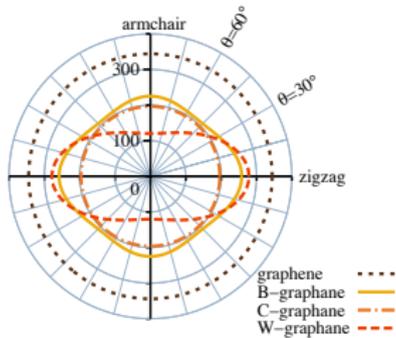
where:  $\Delta = C_{11}C_{22} - C_{12}^2$        $c = \cos \theta$        $s = \sin \theta$

## 2 Trigonal symmetry

- obtained by imposing the symmetry condition  $C_{11} = C_{22}$  and the Cauchy relation  $2C_{44} = C_{11} - C_{12}$

### Key features

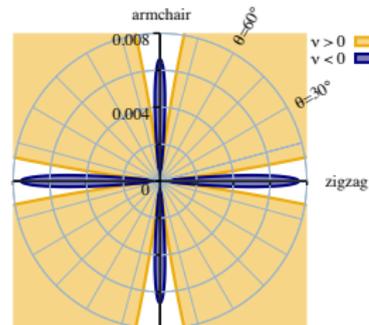
- 1  $E_{\vec{n}}$  and  $\nu_{\vec{n}}$  directly obtained by the linear elastic constants  $C_{ij}$   
 → no need to mimic a traction experiment along the arbitrary direction identified by  $\vec{n}$  or  $\theta$
- 2 in-plane homogeneous deformations
  - **isotropic case:** (i) an axial deformation along the zigzag direction; and (ii) an hydrostatic planar deformation
  - **anisotropic case:** (iii) an axial deformation along the armchair direction; and (iv) a shear deformation



**MEMO**  
**circular shape:**  
**fully isotropic behavior**

Unconventional behavior of  
 B-graphane

- Poisson ratio vanishingly small along  $zz$  and  $ac$
- actually, **Poisson is negative !**





## Nonlinear constitutive equation

$$\sigma_{\bar{n}} = E_{\bar{n}}\epsilon_{\bar{n}} + D_{\bar{n}}\epsilon_{\bar{n}}^2$$

- 1  $D_{\bar{n}}$  is a complicated expression of  $C_{ijk}$
- 2 **orthogonal symmetry:**

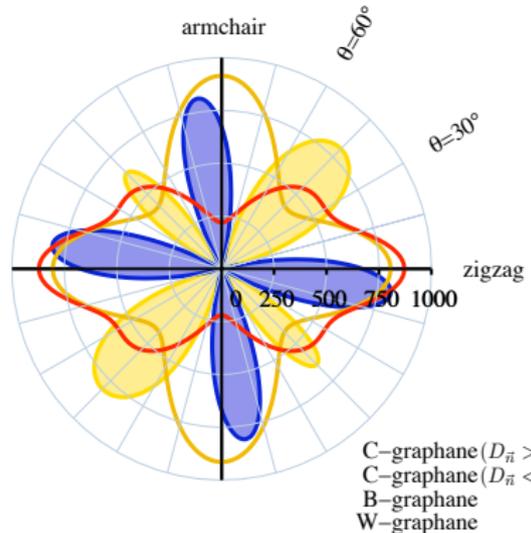
$$\text{all } C_{ijk} < 0 \rightarrow D_{\bar{n}}^{(ortho)} < 0$$

both B- and W-graphane show an  
**hyperelastic softening behavior**

- 3 **trigonal symmetry:**

$$C_{114}, C_{144} > 0$$

C-graphane shows both  
**hyperelastic hardening**  
 or  
**hyperelastic softening behavior**



## Conclusions - *Cadelano, Palla, Giordano, Colombo, PRB 82, 235414 (2010)*

- 1 **All graphane conformers** respond to any arbitrarily-oriented extension with a **much smaller lateral contraction** than graphene
- 2 **B-graphane** has a **small and negative Poisson ratio** along  $zz$  and  $ac$  directions  
→ **axially auxetic elastic behavior**
- 3 **C-graphane** admits **both softening and hardening hyperelasticity**

## Next steps

- disordered systems & H-motifs: tailoring elastic moduli through H-decorantion (?)
- thermo-elasticity

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