

Elastic Properties of Hydrogenated Graphene

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in collaboration with

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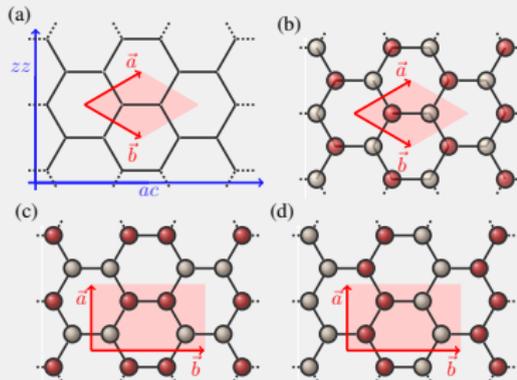


- 1 Introduction
 - Basic definitions
 - Hybridization vs. properties
 - This work
- 2 Theoretical set-up
- 3 Structure and stability
 - Structure
 - Stability
- 4 Calculation of elastic constants
 - Method
 - Validation
 - Second-order stiffness constants (linear elasticity)
- 5 Mechanical behavior under loading
 - Linear regime
 - Nonlinear elasticity
- 6 Conclusions

Graphane is the hydrogenated form of graphene

- a two-dimensional, periodic, and covalently bonded **hydrocarbon with a C:H ratio of 1**
- hydrogen atoms decorate the carbon honeycomb lattice on **both the top and bottom side**
- any graphane conformer retains **high crystallographic symmetry** (either trigonal or orthorhombic)

Possible hydrogen decorations



Red: top hydrogen atoms
Grey: bottom hydrogen atoms

Shaded areas: unit cell with lattice vectors

(a): graphene scaffold (full lines) with zigzag (zz) and armchair (ac) directions

(b, c, d): **chair-**, **boat-**, and **washboard-graphane**

- **Graphane**

at first theoretically predicted

Phys. Rev. B **75**, 153401 (2007) - *Phys. Rev. B* **77**, 035427 (2008)

eventually grown

Science **323**, 610 (2009)

since then, many more papers...

- **Attractive feature**

hydrogenation: $sp^2 \rightarrow sp^3$ change in orbital hybridization \rightarrow new properties

- **Change in the orbital hybridization**

- ① it has been calculated that graphane is an insulator, with an energy gap as large as ~ 3 eV
Phys. Rev. B **75**, 153401 (2007) - *Phys. Rev. B* **77**, 035427 (2008)
- ② partially H-covered and disordered samples show unlike electronic and phonon properties than graphene
Science **323**, 610 (2009)
- ③ calculated in-plane stiffness and Poisson ratio of graphane are smaller than those of graphene
- ④ yield strain is predicted to vary upon temperature and stoichiometry
Topsakal et al., Appl. Phys. Lett. **96**, 091912 (2010)

Our perspective: sp^2 -to- sp^3 change in orbital hybridization is expected to cause

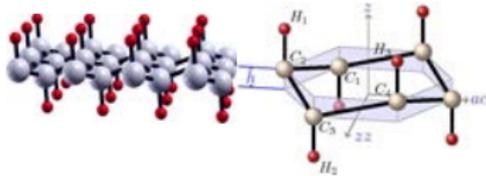
major variations in the mechanical properties of hydrogenated graphene

Highlights

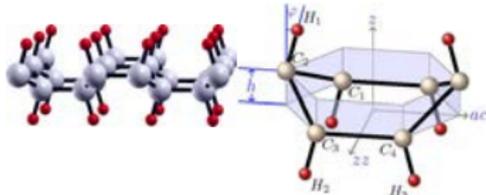
- 1 characterize non-isotropic linear elastic features
contrary to graphene: Science 321, 385 (2008) - Phys. Rev. Lett. 102, 235502 (2009)
- 2 increased or reduced “rigidity” ?
linear elastic moduli most likely affected
- 3 investigate overall mechanical behavior of the systems
anisotropic dependence of its response to external loads

Approach: blending together atomistic and continuum modeling

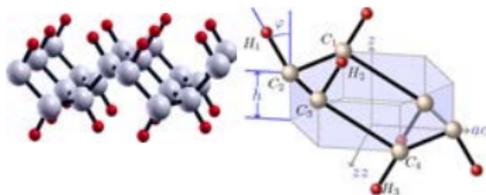
- **Step 0** - Search for equilibrium structures of graphene conformers and establish their stability
- **Step 1** - Calculate 2nd-order C_{ij} and 3rd-order C_{ijk} elastic constants
- **Step 2** - Elaborate a continuum description for (anisotropic) elastic moduli
- **Step 3** - Investigate mechanical behavior upon loading



(a) C-graphane



(b) B-graphane



(c) W-graphane

Graphane conformers have very different structures

- 1 **specific H sublattice and different buckling**
 - C-graphane: H alternate on both sides
 - B-graphane: H alternate along ac
 - W-graphane: H along z double rows on both sides

- 2 **C-C bond length**
 - C-graphane and W-graphane: similar to sp^3 systems and much larger than in graphene → bond length 1.54\AA
 - B-graphane: two types of C-C bonds
 - ◇ connecting two C atoms lying on opposite sides → bond length 1.57\AA
 - ◇ connecting two C atoms lying on same sides → bond length 1.54\AA

- 3 **C-H bond length**
 typical of any hydrocarbon: 1.1\AA

Phonons* and energetics

1 phonon branches

- no soft modes (i.e. no instabilities)
- linear k -dependence for LA and TA modes
- k^2 -dependence for flexural (ZA) mode

2 anisotropic elastic behavior

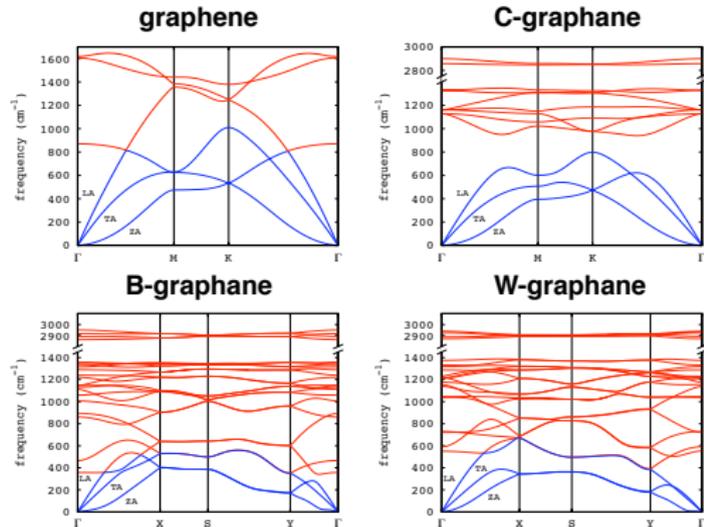
- C-graphane: isotropic speed of sound
- B- and W-graphane: different sound velocities

3 energetics

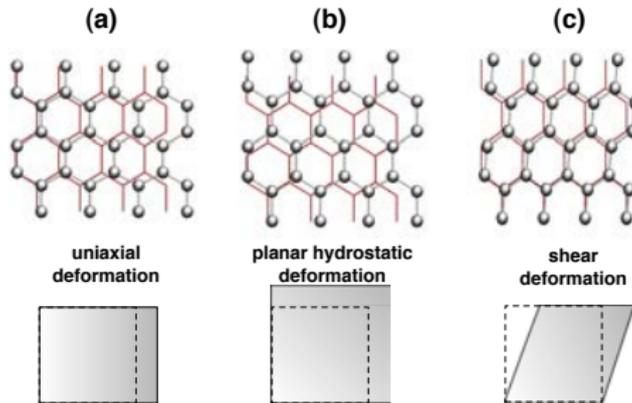
- C-graphane: energetically favored conformer
- W- and B-graphane: higher GS energy
 0.05 and 0.10 eV per C-H unit
- all conformers thermodynamically accessible,
 as indeed experimentally guessed

* Compare well with

- ◇ *Phys. Rev. Lett.* **105**, 037002 (2010) for C-graphane
- ◇ *arXiv:1011.0018v1 [cond-mat.mtrl-sci]*



- Total-energy calculations** performed by Density Functional Theory (DFT)
QUANTUM ESPRESSO *package* - *J. Phys.: Condens. Matter* **21** 395502 (2009)
 - GGA-like XC functional - Vanderbilt ultrasoft pseudopotential
 - 50Ry energy cutoff - (18x18x3) Monkhorst-Pack grid for BZ sampling
- Atomic positions optimized** by using the quasi-Newton algorithm and PBCs
interactions between adjacent atomic sheets hindered by a large spacing $> 10 \text{ \AA}$
- Elastic moduli obtained from the (energy) vs. (in-plane strain) curves $U = U(\zeta)$**



$$U(\zeta) = U_0 + \frac{1}{2}U^{(2)}\zeta^2 + \frac{1}{6}U^{(3)}\zeta^3 + O(\zeta^4)$$

Benchmark calculations

- energy-vs-strain obtained by in-plane deformations through very many data-points
a technically important issue!
- in-plane deformations defined by a single parameter ζ
- deformations applied up to 5%

		Young [Nm^{-1}]	Poisson
graphene	this work	344	0.169
	PRB 64 , 235406 (2001)	345	0.149
	PRB 76 , 064120 (2007)	350	0.186
C-graphane	this work	246	0.080
	APL, 96 , 091912 (2010)	243	0.070

	graphene	C-	graphane	W-
		C-	B-	W-
C_{11}	354	248	258	280
C_{22}			225	121
C_{12}	60	20	-1.7	14
C_{44}	147(*)	114(*)	93	81

units of Nm^{-1} (*) $2C_{44} = C_{11} - C_{12}$

Qualitative trends

- 1 $C_{11} - C_{12}$ difference
 → **W-graphane the most elastically anisotropic conformer**
- 2 C_{44} modulus
 → **resistance to a shear deformation decreases monotonically**
- 3 C_{12} modulus (or, similarly, the Poisson ratio)
 → much smaller than in pristine graphene
 → **lateral contraction upon extension affected by the new hybridization**
- 4 **B-conformer characterized by a negative C_{12} value**

- ① **C-graphane**: trigonal symmetry (i.e. elastically isotropic as hexagonal graphene)

$$U_{trigo} = \frac{1}{2}C_{11} (\epsilon_{xx}^2 + \epsilon_{yy}^2 + 2\epsilon_{xy}^2) + C_{12} (\epsilon_{xx}\epsilon_{yy} - \epsilon_{xy}^2)$$

- ② **B- and W-conformers**: orthorhombic symmetry (i.e. anisotropic linear elastic)

$$U_{ortho} = \frac{1}{2}C_{11}\epsilon_{xx}^2 + \frac{1}{2}C_{22}\epsilon_{yy}^2 + C_{12}\epsilon_{xx}\epsilon_{yy} + 2C_{44}\epsilon_{xy}^2$$

- ③ **constitutive in-plane stress-strain equation** are straightforwardly

$$\hat{T} = \frac{\partial U}{\partial \hat{\epsilon}} \rightarrow \begin{cases} T_{xx} &= C_{11}\epsilon_{xx} + C_{12}\epsilon_{yy} \\ T_{yy} &= C_{22}\epsilon_{yy} + C_{12}\epsilon_{xx} \\ T_{xy} &= 2C_{44}\epsilon_{xy} \end{cases}$$

Procedure to get elastic moduli

- **apply an axial tension** σ along the arbitrary direction $\vec{n} = \cos \theta \vec{e}_x + \sin \theta \vec{e}_y$
 → θ is the angle between applied tension and zigzag direction
- **get stress components**: $T_{xx} = \sigma \cos^2 \theta$ $T_{yy} = \sigma \cos \theta \sin \theta$ $T_{yy} = \sigma \sin^2 \theta$
- invert the constitutive equation → get **longitudinal strain** ϵ_l **transverse strain** ϵ_t
- **elastic moduli**: $E_{\vec{n}} = \frac{\sigma}{\epsilon_l}$ $\nu_{\vec{n}} = -\frac{\epsilon_t}{\epsilon_l}$

1 Orthogonal symmetry

- Young modulus

$$E_{\vec{n}} = \frac{\Delta}{C_{11}s^4 + C_{22}c^4 + \left(\frac{\Delta}{C_{44}} - 2C_{12}\right)c^2s^2}$$

- Poisson ratio

$$\nu_{\vec{n}} = -\frac{\left(C_{11} + C_{22} - \frac{\Delta}{C_{44}}\right)c^2s^2 - C_{12}(c^4 + s^4)}{C_{11}s^4 + C_{22}c^4 + \left(\frac{\Delta}{C_{44}} - 2C_{12}\right)c^2s^2}$$

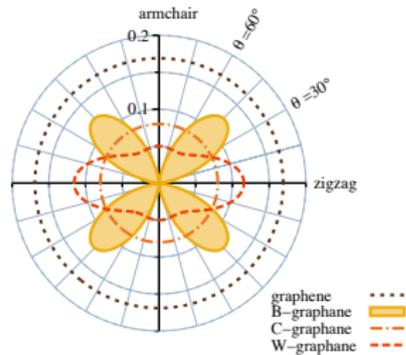
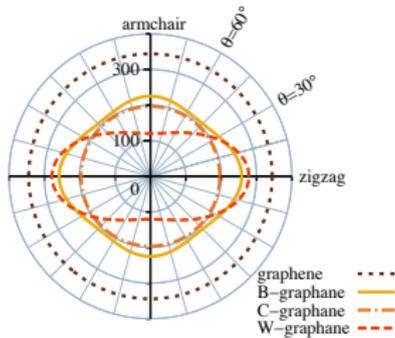
where: $\Delta = C_{11}C_{22} - C_{12}^2$ $c = \cos \theta$ $s = \sin \theta$

2 Trigonal symmetry

- obtained by imposing the symmetry condition $C_{11} = C_{22}$ and the Cauchy relation $2C_{44} = C_{11} - C_{12}$

Key features

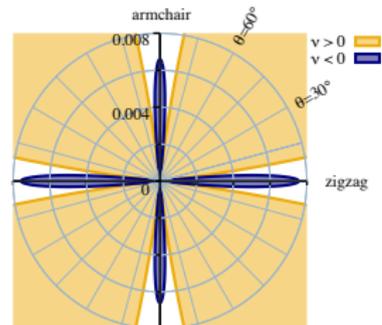
- $E_{\vec{n}}$ and $\nu_{\vec{n}}$ directly obtained by the linear elastic constants C_{ij}
 → no need to mimic a traction experiment along the arbitrary direction identified by \vec{n} or θ
- in-plane homogeneous deformations
 - isotropic case:** (i) an axial deformation along the zigzag direction; and (ii) an hydrostatic planar deformation
 - anisotropic case:** (iii) an axial deformation along the armchair direction; and (iv) a shear deformation



MEMO
circular shape:
fully isotropic behavior

Unconventional behavior of
 B-graphane

- Poisson ratio vanishingly small along zz and ac
- actually, **Poisson is negative !**



Nonlinear constitutive equation

$$\sigma_{\bar{n}} = E_{\bar{n}}\epsilon_{\bar{n}} + D_{\bar{n}}\epsilon_{\bar{n}}^2$$

- 1 $D_{\bar{n}}$ is a complicated expression of C_{ijk}
- 2 **orthogonal symmetry:**

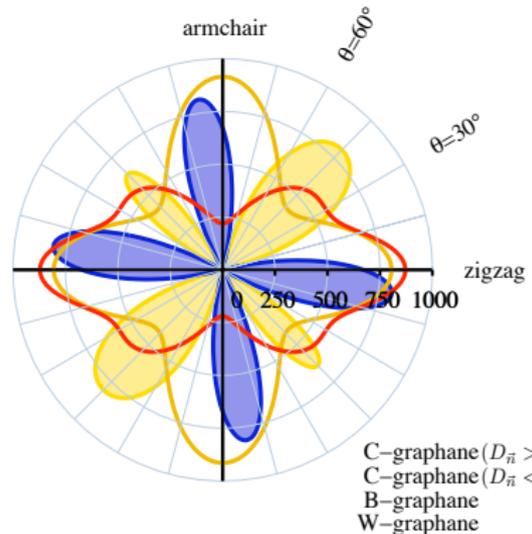
$$\text{all } C_{ijk} < 0 \rightarrow D_{\bar{n}}^{(ortho)} < 0$$

both B- and W-graphane show an
hyperelastic softening behavior

- 3 **trigonal symmetry:**

$$C_{114}, C_{144} > 0$$

C-graphane shows both
hyperelastic hardening
 or
hyperelastic softening behavior



Conclusions - *Cadelano, Palla, Giordano, Colombo, PRB 82, 235414 (2010)*

- 1 **All graphane conformers** respond to any arbitrarily-oriented extension with a **much smaller lateral contraction** than graphene
- 2 **B-graphane** has a **small and negative Poisson ratio** along zz and ac directions
→ **axially auxetic elastic behavior**
- 3 **C-graphane** admits **both softening and hardening hyperelasticity**

Next steps

- disordered systems & H-motifs: tailoring elastic moduli through H-decorantion (?)
- thermo-elasticity

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